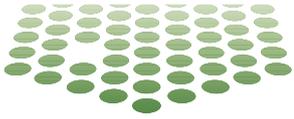


Gels in crystallization of biological macromolecules: additives with numerous properties

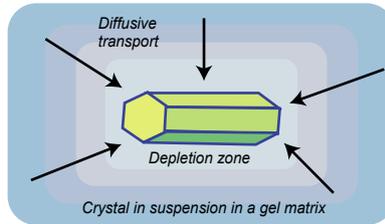


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Gels and crystallization

Gels have been known for long to promote the growth of high-quality crystals of small molecules. During the last decade different types of gels - agarose, silica or polyacrylamide gels - have been tested on the crystallization of biological macromolecules. They are still seldom used, although they provide many advantages from the crystal production to the crystal handling [1].

When a crystal is growing, it takes up and incorporates molecules from the surrounding solution. Thus, the crystal creates a *depletion zone* in the mother liquor. In usual conditions, the resulting concentration gradient and the mass transport are permanently disturbed by convection currents. On the other hand, when a gel is introduced in the crystallization medium, the convection is reduced and the molecule transport towards the crystal is more regular and operates in a diffusive mode.



This leads to a crystalline material of improved quality. Indeed, recent comparative studies have shown that crystals grown in gel present a lower mosaicity and produce a more intense diffraction signal [2,3].

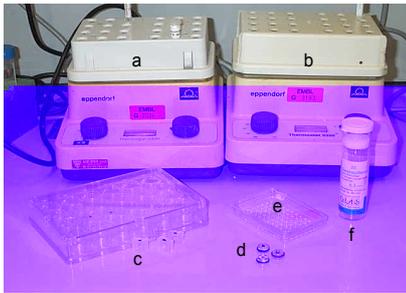
Other practical advantages of gels [1]:

- they keep the crystals in suspension in their mother liquor
- they prevent sedimentation
- they promote a growth in 3 dimensions
- they provide a mechanical protection (during soaking, mounting, transport...)

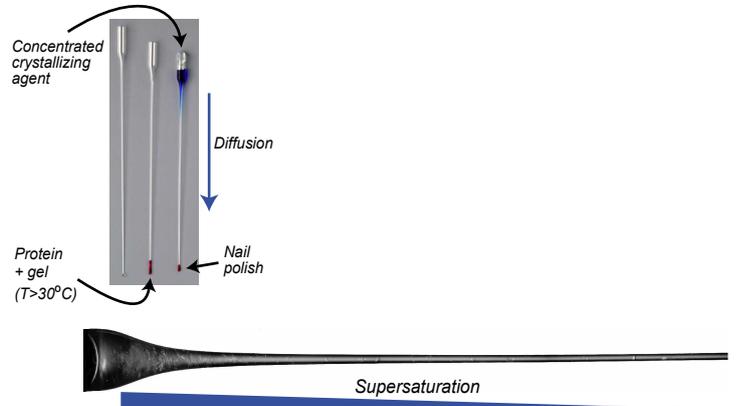
In addition gels are compatible with most common crystallization methods and offer new possibilities like counter-diffusion in capillaries.

Crystallization in agarose gel

Agarose is a very convenient gel to work with. It is even easier when using a gel with a low gelling temperature (~28°C) which is more compatible with the presence of heat-sensitive macromolecules. A 2% (w/v) agarose stock solution is stored at 4°C. Before setting-up crystallization trials, this solution is melted at 85-90°C (a) and kept at 35-45°C (b). Then it is added to the crystallizing agent or to the macromolecule solution to a final concentration of 0.1-0.2% (w/v). This mixture can either be used in vapor diffusion (c), dialysis (d) or batch (e) assays, and furthermore it can be utilized to perform counter-diffusion experiments in capillaries (f).

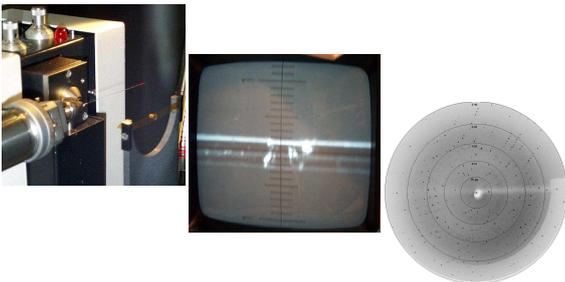


The counter-diffusion method



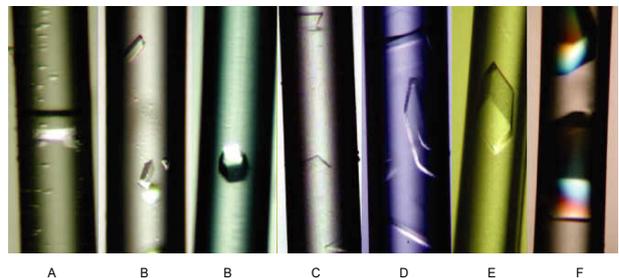
Crystal analysis

Because of the low agarose gel concentration (0.1-0.2 % (w/v)), crystals are easily recovered before mounting in capillaries or in cryo loops. The presence of the gel can even facilitate flash-freezing [7]. As shown below for lysozyme, crystals that were grown in capillaries using the counter-diffusion method can be analyzed *in situ* at room temperature or in cryogenic conditions [6].



A few examples

A	Host Factor I (HF-I)	<input type="checkbox"/>	<input type="checkbox"/>	<i>Escherichia coli</i>	<input type="checkbox"/>	66 kDa (hexamer)
B	Sm-like protein (Sm1)	<input type="checkbox"/>	<input type="checkbox"/>	<i>Aeropyrum pernix</i>	<input type="checkbox"/>	59 kDa (heptamer)
C	Endonuclease VII (EndoVII)	<input type="checkbox"/>	<input type="checkbox"/>	Bacteriophage T4	<input type="checkbox"/>	36 kDa (dimer)
D	EndoVII / DNA cruciform junction	<input type="checkbox"/>	<input type="checkbox"/>	"	<input type="checkbox"/>	36 / 13 kDa
E	EndoVII / mismatched DNA duplex	<input type="checkbox"/>	<input type="checkbox"/>	"	<input type="checkbox"/>	36 / 13 kDa
F	Lysozyme	<input type="checkbox"/>	<input type="checkbox"/>	<i>Gallus gallus</i>	<input type="checkbox"/>	14 kDa (monomer)



Capillary diameter: 0.3 mm

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Conclusion ...

Crystallization in gel

- is easy to perform
- can be adapted to all current crystallization methods
- makes the use of counter-diffusion techniques possible
- provides better growth conditions
- leads to crystals of enhanced quality

You should definitely try it!